

Physical Properties and Light-Related Applications of Black Silicon Structures

Emil Pincik^{1*}, Hikaru Kobayashi², Róbert Brunner¹, Kentaro Imamura², Milan Mikula³, Michal Kucera⁴, Pavel Vojtek⁵, Zuzana Zabudla⁵, PeterSvec Sr¹, Ján Gregus⁵ and PeterSvec Jr¹

1. Institute of Physics, SAS, Dubravska cesta 9, 845 11 Bratislava, Slovak Republic

2. Institute of Scientific and Industrial Research, Osaka University and CREST, 8-1 Mihogaoka, Ibaraki, Osaka 567-0047, Japan

3. Faculty of Chemical and Food Technology of STU, Radlinského 9, 812 37 Bratislava, Slovak Republic

4. Institute of Electrical Engineering SAS, Dubravska cesta 9, 841 04 Bratislava, Slovak Republic

5. Faculty of Mathematics, Physics, and Informatics of Comenius University, Mlynska Dolina, F2, 842 48 Bratislava, Slovak Republic

Abstract: This contribution deals with the black silicon (BS) nanocrystalline specimens produced using the surface structure chemical transfer method (SSCT). This method can produce a nanocrystalline Si black color layer on c-Si with a thickness range of ~50 nm to ~300 nm via the contact of c-Si immersed in the chemical solution $HF + H_2O_2$ with a catalytic mesh. The photoluminescence properties are related to the formation of nanocrystals, the structural properties of which are similar to those formed on the back of a sawn Si wafer and the resulting splitting of large Si crystals. X-ray diffraction of the Si front and back sides confirms the dominant reflection of the 311 Si crystalline planes. We suppose that the formation of the black silicon over-layer is pre-determined by the crystalline defects induced by the applied sawing procedure, even though saw damage (defects introduced by the applied sawing procedure) is not necessary for the SSCT method. The formation of the pn type Si solar cell is presented, including black silicon over-layer and without antireflection coating, with efficiency of ~19.1%.

Key words: Photoluminescence, black silicon, solar cell.

1. Introduction

Solar cell structures need low reflectance surfaces to maximize the number of absorbed incident photons, with antireflection (AR) coatings commonly being used in the solar cell industry. However, the use of such coatings is limited because they reduce reflection for a narrow range of light wavelengths [1, 2]. Jansen et al. [3] firstly demonstrated the replacement of conventional AR coatings with a black silicon layer. Nanocrystalline black Si samples can be produced using multiple methods [4, 5].

Nanocrystalline black Si is formed by an electrochemical reaction from the surface, with a nanocrystalline Si layer forming on the surface. The used chemical solution penetrates the space region of

the nanocrystalline Si layer, leading to the growth of the nanocrystalline Si layer. With this formation mechanism a low reflectivity nanocrystalline Si layer is formed. The low reflectivity is probably caused by the special depth profile of the refractive index, which is close to 1 near the surface and increases with depth. We suppose that the space/Si volume ratio in the nanocrystalline Si layer is high near the surface and decreases with depth. In the case of the multicolor wafer, the space/Si volume ratio seems to be low even near the surface, leading to relatively high reflectivity of ~10%.

Si nanocrystallites embedded in SiO_2 thin films were investigated by Ding et al. [6]. Observed photoluminescence (PL) spectra demonstrated a continuous annealing temperature-based evolution. In dependence on the annealing temperature, six PL bands located at 415, 460, 520, 630, 760 and 845 nm,

^{*}**Corresponding author:** Emil Pincik, PhD., research fields: development of solar cells, semiconductor physics and technology.

respectively, have been observed, whereby annealing at 1,100 °C yields the strongest PL band at 760 nm. Wong et al. [7] observed significantly enhanced PL bands at 650 nm and 750 nm in a similar structure (Si nanocrystals in SiO₂) after high temperature annealing at 1,100 °C. They explained this effect by the formation of Si nanocrystallites via a phase separation reaction, and by the removal of defect-related absorption centers during annealing. Kuznetsov et al. [8] investigated the PL properties of an array of vertically aligned Si nanopillars with average diameters of ~100 nm and ~5 µm length. A strong broad visible PL emission band around 1.8 eV, with a nearly perfect Gaussian shape, was observed in porous-walled nanopillars - Si nanocrystals passivated with $SiO_{(x)}$. The authors found the Si nanocrystal surface most probably to be responsible for the luminescence. The red shift of the emission maximum and the luminescence quenching induced by oxidation of the samples in the UV ozone partly confirmed this assumption. A model of luminescence has been proposed involving UV photon absorption by Si nanocrystals with a subsequent exciton radiative recombination on defect sites in $SiO_{(x)}$ covering the Si nanocrystals.

Lü et al. [9] observed a PL band with a maximum in the range 0.78-0.84 eV on black Si structures when the annealing temperature changed from 500 °C to 1,200 °C, respectively. Goshi et al. [10] investigated the origin of the visible and near infrared PL of chemically etched Si nanowires adorned with arbitrarily shaped Si nanocrystals. Their studies revealed that Si nanocrystals are primarily responsible for the emission in the 1.5-2.2 eV range, in dependence on the cross-section of the Si nanocrystals. The large diameter Si/SiO(x) nanowires yield a distinct NIR PL signal that consists of peaks at 1.07, 1.10 and 1.12 eV. The latter NIR peaks were attributed to the room temperature TO/LO phonon-assisted radiative recombination of free carriers condensed in the electron-hole plasma in etched Si nanowires. The enhancement of PL intensity and the red shift of the PL peak measured at low temperatures were explained as a consequence of the interplay of radiative and non-radiative recombinations at the interfaces of Si nanocrystals and Si/SiO(x) nanowires. Researchers at ISIR Osaka University developed a fabrication method of low reflectivity Si surfaces in which a mold with catalytic Pt layer is placedinto contact with Si immersed in a $H_2O_2 + HF$ solution [11]. The energy of PL maximum observed for the black silicon layer prepared on a polycrystalline Si wafer at room temperature was ~1.85 eV [12]. A similar method was also applied for the preparation of the OBSi samples in this contribution. An additional alternative technological approach was developed in the same laboratories in Japan by M. Takahashi et al. [13].

In this contribution we will present and discuss the optical properties of both a black silicon sample and the front and back sides of the reference Si sample (without RCA treatment), respectively, and the properties of solar cell structures prepared both with and without black silicon over-layer.

2. Materials

Black silicon nanocrystalline specimens were produced sing the surface structure chemical transfer method. This method can produce a nanocrystalline Si layer on c-Si with a thickness of ~50 nm (multi-color) to ~300 nm (black color) by the contact of c-Si immersed in chemical solutions of HF + H_2O_2 with a catalytic mesh.

In the case of the SSCT method, laboratories of the ISIR Osaka University in Japan can produce ultralow reflectivity surfaces using mirror-like polished Si wafers (for LIS use), and therefore, defects are not necessary for the production of a nanocrystalline Si layer. The SSCT method forms a nanocrystalline Si layer with the following electrochemical reaction:

Decomposition of H_2O_2 on Pt to generate positive charges:

$$H_2O_2 + H^+ \to H_2O + OH^- + 2h \tag{1}$$

Positive charges are injected to Si, diffuse to Si surfaces, and holes (h) are captured at surface states, e.g., Si dangling bond states. In the presence of holes, Si strongly reacts with HF, resulting in dissolution of Si.

$$Si + 6HF \rightarrow H_2 SiF_6 + H_2 + 2H^+ \qquad (2)$$

surface Because states present are inhomogeneously, nonuniform Si dissolution occurs, resulting in the formation of the nanocrystalline Si layer. Acid etching used for texturing of polycrystalline Si wafers proceeds from saw damage, and therefore, on polycrystalline Si wafers produced using the fixed abrasive technology without high density saw damage defects, acid etching cannot form textured structure while the SSCT method can form ultralow reflectivity surface from polycrystalline Si wafers produced with the fixed abrasive technology. This is another evidence that saw damage (defects introduced by the applied sawing procedure) is not necessary for the SSCT method.

All black silicon samples were prepared on p-type (100) Si substrates after RCA cleaning, and were stabilized with thermal oxidation at approx. 900 C for a few minutes. Hence the abbreviation OBSi is used. The optical properties of the reference Si sample after polishing procedures and brief etching in only vapor HF are also investigated. The corresponding abbreviation is Ref Si.

The form of catalytic mesh and the technology of OBSinano-structure preparation are illustrated in Fig. 1.

2.1. Used Experimental Methods

2.1.1 Leica DM2700

Standard optical microscope was used to determine the morphology of treated surfaces and/or back sides of the silicon structures. The AFM records were obtained with VEECO CP2 equipment. Tip radius of 10 nm and a contact force of 10 nN were used in contact mode.

2.1.2 Digilab Excalibur FTS 3000 MX

The spectrometer was used for FTIR spectroscopy

with the following FTIR spectroscopy software: ResolutionsPro 5.2.0, Agilent with Kramers-Kronig transformation. A PIKE Technologies Specular Reflectance Accessory was used at an angle of 30° . The measured area's diameter was 10 mm, used resolution: 4 cm⁻¹, number of scans: 60.

2.1.3 Bruker D8 Advance

X-ray diffraction at grazing incidence angles has been measured with this equipment, with a 60 mm Goebel mirror for Cu radiation and 5° Soller slit in the primary beam path. The secondary beam path was fitted with a 0.23° parallel plate collimator and LiF monochromator to enhance instrument resolution. The emitted beam shape was 10 mm \times 1.2 mm.

2.1.4 Photoluminescence (PL)

The measurements were performed at room and 6 K temperatures with equipment that used a halogen lamp as the light source, and a PL spectrum in the range 420-1,000 nm was detected by the CCD element.

3. Results and Discussion

3.1 Optical and X-Ray Methods

The photoluminescence records of Fig. 2 were recorded on both Ref Si (surface side after polishing and cleaning with HF vapor) and OBSi, respectively, with PL apparatus with a focusing system and cooled Si photodetector. These records illustrate the blue shift of the PL spectra by approx. 0.40 eV, with the change of the sample temperature from room one to 6K. During the PL measurements, the OBSi sample was hung outside the metallic sample holder to avoid any potential influence of the supporting material on the PL spectra form. The results of fitting procedures are



Fig. 1 The form of catalytic mesh and the technology of OBSinano-structure preparation.



Fig. 2 Photoluminescence signals recorded at room temperature and 6 K on both Ref Si (surface side after polishing and cleaning with HF vapor) and the OBSi.

presented in Table 1, Table 2 and Table 3. Three peaks fitting Gaussian curves have been optimal for these structures due to the lowest value of residuum. The corresponding analytical procedure has been described in detail in Ref. [14].

On the top and back sides of the reference Si sample after the polishing procedures and HF etching (removing SiO₂ native oxide), photoluminescence signals were also recorded at room temperature. The PL record of the reference Si back side is shown in Fig. 3. The absolute value of the PL signal was higher than that observed on the OBSi by approximately two orders. The data obtained with the fitting procedure are presented in Table 4. The morphology of the reference Si back side is shown in Fig. 4 obtained with an optical microscope. Fig. 5 shows the surface of the Si back side obtained by the AFM. The roughness of the sample is ~127 nm. Fig. 6 presents the FTIR reflectance spectra of both the front and back side of the reference Si sample obtained with a diamond tip. Dispersion dependences of n and k on the wavenumber of both investigated Ref Si regions (front and back sides) were calculated from the IR reflectance measurements at 30° – see Figs. 7 and 8, respectively. Standard commercial software was used. Prior to RCA cleaning, the thickness of the structurally disturbed Si surface region resting after the

Table 1 Data obtained by fitting procedures of PL records of the front side of the Si reference sample at room temperature (a) and at 6 K (b).

		(a)	
No.	Energy (eV)	Intensity (A.U.)	FWHM (eV)
1	1.57	7.76e-4	0.11
2	1.70	1.04e-3	0.24
3	1.89	7.53e-4	0.40
Offset	2.68e-5	(A.U.)	
Residuum	1.58e-3	(A.U.)	
		(b)	
No.	Energy (eV)	Intensity (A.U.)	FWHM (eV)
No.	Energy (eV) 1.91	Intensity (A.U.) 3.38e-3	FWHM (eV) 0.45
No. 1 2	Energy (eV) 1.91 2.10	Intensity (A.U.) 3.38e-3 4.95e-3	FWHM (eV) 0.45 0.29
No. 1 2 3	Energy (eV) 1.91 2.10 2.27	Intensity (A.U.) 3.38e-3 4.95e-3 3.11e-3	FWHM (eV) 0.45 0.29 0.19
No. 1 2 3 Offset	Energy (eV) 1.91 2.10 2.27 0	Intensity (A.U.) 3.38e-3 4.95e-3 3.11e-3 (A.U.)	FWHM (eV) 0.45 0.29 0.19

Table 2Data obtained by fitting procedures of PL recordsof the front side of the OBSi at room temperature (a) and at6 K (b).

		(a)	
No.	Energy (eV)	Intensity (A.U.)	FWHM (eV)
1	1.57	1.44e-3	0.11
2	1.71	2.48e-3	0.24
3	1.88	2.13e-3	0.37
Offset	2.56e-5	(A.U.)	
Residuum	2.93e-3	(A.U.)	
-			
		(b)	
No.	Energy (eV)	(b) Intensity (A.U.)	FWHM (eV)
No.	Energy (eV) 1.86	(b) Intensity (A.U.) 8.78e-4	FWHM (eV) 0.44
No. 1 2	Energy (eV) 1.86 2.09	(b) Intensity (A.U.) 8.78e-4 2.21e-3	FWHM (eV) 0.44 0.29
No. 1 2 3	Energy (eV) 1.86 2.09 2.24	(b) Intensity (A.U.) 8.78e-4 2.21e-3 1.18e-3	FWHM (eV) 0.44 0.29 0.19
No. 1 2 3 Offset	Energy (eV) 1.86 2.09 2.24 0	(b) Intensity (A.U.) 8.78e-4 2.21e-3 1.18e-3 (A.U.)	FWHM (eV) 0.44 0.29 0.19

Table 3Values of the blue shift of fittedphotoluminescence spectra with the change of the sampletemperature from room one to 6 K. The OBSi sample hasbeen hanged near the sample support.

(a)

$\mathbf{D} = \mathbf{f} \mathbf{I} \left(\mathbf{C} \mathbf{V} \right)$	D of I (DT)	

Kel I (6 I	\mathbf{x}) - Kel I (KI)		
No.	Energy (eV)	Energy (eV)	Energy (eV)
1	1.91	1.57	0.34
2	2.10	1.70	0.40
3	2.27	1.89	0.38
		(b)	
OBSi (6	K) - OBSi (RT)		
No.	Energy (eV)	Energy (eV)	Energy (eV)
1	1.86	1.57	0.29
2	2.09	1.71	0.38
3	2.24	1.88	0.36



Fig. 3 Photoluminescence of the reference Si back side recorded at room temperature.

Table 4Data obtained by fitting procedures of PL recordsrecorded at room temperature of the back side of the Sireference sample.

No.	Energy (eV)	Intensity (A.U.)	FWHM (eV)
1	1.56	1.45e-1	0.10
2	1.67	2.00e-1	0.20
3	1.82	3.12e-4	0.32
Offset	7.41e-4	(A.U.)	
Residuum	3.14e-1	(A.U.)	



Fig. 4 The morphology of the reference Si back side obtained with optical microscope.

polishing procedure is ~100 nm. This damaged region also contains broken surface parts of the c-Si wafer of varying sizes covered with native $SiO_{(x)}$. The Si back side contains larger residues of crystalline parts of the original Si crystal. Fig. 9 shows the X-ray diffraction pattern of the Ref Si back side recorded at the following three grazing incidence angles -1° , 2° and 3° . Reflection 311 dominates. A comparison of the XRDGI patterns of Fig. 9 and those presented in [15]



Fig. 5 The surface of the Si back side obtained by AFM.



Fig. 6 The FTIR reflectance spectra of both the front and back side of the reference Si sample obtained with a diamond tip.



Fig. 7 Dispersion dependences of *n* on the wavenumber of the front and back sides of the reference Si after polishing procedures.



Fig. 8 Dispersion dependences of *k* on the wavenumber of the front and back sides of the reference Si after polishing procedures.



Fig. 9 X-ray diffraction patterns of the Ref Si back side recorded at the following three grazing incidence angles -1° , 2° and 3° . Reflection 311 dominates.

confirm that the XRDGI records of all structures (OBSi, Ref Si front and back sides) are very similar, in fact almost identical. The dominant difference is given by the absolute value of structures' intensities. In comparison with the results obtained on the porous Si and published in [16], a stronger signal of reflection 111 is missing in the case of all investigated samples.

One of the determining parameters influencing the absolute intensity value of the recorded PL signals in excited optical regions of the front and back sides of the Ref Si sample can be related to: i) the illuminated size of the sample areas, and ii) the nanocrystalline structure of the illuminated Si side.

Recorded PL signals are related to the formation of nanocrystals, the structural properties of which are similar to those formed on the back side of the Si wafer after the large Si crystal has been sawn. X-ray diffraction of the Si front and back sides confirms the dominant reflection of the 311 Si crystalline planes. We suppose that the formation of the black silicon over-layer is pre-determined by crystalline

Recorded PL signals are related to the formation of nanocrystals the structural properties of which are similar to those formed on the back side of the Si wafer after the large Si crystal has been sawn. X-ray diffraction of the Si front and back sides confirms the dominant reflection of the 311 Si crystalline planes. We suppose that the formation of the black silicon over-layer is pre-determined by crystalline defects induced by the used sawing procedure. Both reference Si sides have very similar PL spectra, even after their deconvolution to three separate Gaussians.

It needs to be stated that correct photoluminescence (PL) signals in the photon energy region 1.3-2.5 eV were not recorded in the polished Si structure after RCA cleaning.

The blue shift of both the PL spectra (reaching almost a value of 0.40 eV (see Table 3) observed after the decrease of sample temperature to approx. 6 K, we also relate to the change of semiconductor band gap width, i.e. with the change of semiconductor properties of structures generating PL light, with the amorphous structure containing Si nanocrystalline grains of the OBSi and the front reference Si surface regions on which the additional effect – quantum confinement – is probably observed. The band gap width of semiconductors typically decreases with increasing temperature, see e.g. [17], due to the change of electron-phonon interaction and the expansion of the lattice. We suppose that in the case of a higher concentration of defect states at the

 SiO_2/Si interface also the form of tails of conduction and valence bands can be strongly changed - reduced by the change of sample temperature of room one to approx. 6 K.

In agreement with our conclusions in the paper [15], we suppose that the PL bands recorded at room temperature between 1.70 eV and 1.90 eV on both Ref I Si and OBSi can be related to the defects formed at the $SiO_{(x)}/Si$ interface region of both structures during their preparation. Our results at least partly confirm, the conclusions of Kuznetsov et al. [8] – the model of luminescence involving UV photon absorption by Si nanocrystals with subsequent exciton radiative recombination on defect sites in $SiO_{(x)}$ covering Si nanocrystals can also be proposed.

Of course, it is necessary to state that broad PL records complete because are not of the above-mentioned two limits determined by both the used monochromator and photodetector, respectively. This fact complicates the precise evaluation and discussion of PL results. Very similar PL behaviors recorded at liquid helium (6 K) and room temperatures on both polished crystalline Si and OBSi samples, respectively, indicate the similar origin of recorded PL bands. As the positions of PL maxima of OBSi structures are mainly related to the size of Si nanocrystallites and $SiO_{(x)}$, we therefore suppose that the size of dominant parts of the luminated OBSi nanostructure is pre-determined by the used cutting and polishing Si procedure, and/or the distribution functions of the number of nanocrystallites (formed by cutting and polishing procedures and by the SSCT method) on their size is very similar. It can be supposed that the electrochemical catalytic process of the formation of nanocrystallites using the SSCT method starts along the nanocrystalline grain boundaries residing in the Si surface region after the RCA cleaning procedure.

3.2. Solar Cells

On the Si sample with ~150 nm thickness of black silicon over-layer, a corresponding pn junction was formed – see Fig. 10.

The solar cell parameters (I-V curves) of structures with the BS are compared with a solar cell formed with the same technological steps on the Si substrate without BS over-layer – see Fig. 11. The solar cell parameters of both structures were measured using standard commercial equipment – the results are shown in Table 5.



Fig. 10 Illustration of the pn junction formed on the black silicon structure.



Fig. 11 Solar cell parameters (I-V curves) of structures with the black silicon over-layer are compared with the solar cell formed by the same technological steps on the Si substrate without the over-layer.

Table 5 Solar cell parameters of both structures – with and without black silicon over-layer.

	V _{oc} (V)	$J_{SC} (mA/cm^2)$	FF	Conversion effic. (%)
With SSCT	0.608	40.8	0.771	19.1
Without SSCT	0.601	27.3	0.771	12.7

4. Conclusions

This contribution deals with the photoluminescence of both i) oxidized black silicon (OBSi) nano-crystalline specimens produced by the surface structure chemical transfer method (SSCT), and ii) both sides of a polished reference Si sample, respectively. The similar PL behaviors recorded at liquid helium and room temperatures on both the polished crystalline Si surface and the OBSi samples, respectively, indicate the similar origin/similar sources of the recorded luminescence light. The absolute value of the PL signal was higher than that observed on the OBSi by approximately two orders. As the positions of PL maxima of all structures are mainly related to the size of Si nanocrystallites and SiO(x), then we suppose that the size of the dominant luminated parts of all structures is pre-determined by the used cutting and polishing Si procedure even though saw damage (defects introduced by the applied sawing procedure) is not necessary for the SSCT method, and/or the distribution functions of the number of nanocrystallites (formed by the cutting and polishing procedures and the SSCT method) on their size is very similar. It can be supposed that the electrochemical catalytic process of the formation of nanocrystallites with the SSCT method starts along the Si nanocrystalline grain boundaries that reside in the Si surface region after RCA cleaning. The blue shift of both the PL spectra, almost reaching a value of 0.40 eV and observed after the decrease of the sample temperature to 6K, we relate to the change of semiconductor band gap width, i.e. the change of semiconductor properties of structures that generate the PL light.

Our model of observed photoluminescence phenomena involves UV photon absorption by Si nanocrystals with subsequent exciton radiative recombination on defect sites in $SiO_{(x)}$ covering Si nanocrystals. The model is valid for photoluminescence observed on both the black silicon

structures and the reference Si (front and back sides), respectively.

We have shown that the efficiency of the pn-type Si solar cell with the black silicon overlayer can reach 19.1% without antireflection coating being used.

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References

- Ma, L. L., Zhou, Y. C., Jiang, N., Lu, X., Shao, J. and Lu, W. et al. 2006. "Wide-Band 'Black Silicon' Based on Porous silicon." *Appl. Phys. Lett.* 88: 171907.
- [2] Walheim, S., Schaffer, E., Mlynek, J. and Steiner, U. 1999. "Structure Formation via Polymer Demixing in Spin-Cast Films." *Science* 283: 520-2.
- [3] Jansen, H., de Boer, M., Legtenberg, R. and Elwenspoek, M. 1995. "The Black Silicon Method: A Universal Method for Determining the Parameter Setting of a Fluorine-Based Reactive Ion Etcher in Deep Silicon Trench Etching with Profile control." J. Micromech. Microeng. 5: 115-20.
- [4] Ayon, A. A., Braff, R., Lin, C. C., Sawin, H. H. and Schmidt, M. A. 1999. "Characterization of a Time Multiplexed Inductively Coupled Plasma Etcher." J. Electrochem. Soc. 146: 339-49.
- [5] Wu, C., Crouch, C. H., Zhao, L., Carey, J. E., Younkin, R. and Levinson, J. A. et al. 2001. "Near-Unity Below-Band -Gap Absorption by Microstructured Silicon." *Appl. Phys. Lett.* 78: 1850-2.
- [6] Ding, L., Chen, T. P., Liu, Y., Ng, C. Y., Yang, M. and Wong, J. I. et al. 2008. "Evolution of Photoluminescence Mechanisms of Si+-Implanted SiO₂ Films with Thermal Annealing." J. Nanosci. Nanotechnol. 8: 3555.
- [7] Wong, C. K., Wong, H. and Filip, V. 2009.
 "Photoluminescence of Silicon Nanocrystals Embedded in Silicon Oxide." *J. Nanosci. Nanotechnol.* 9 (Feb.): 1272-6.
- [8] Kuznetsov, A. S., Shimizu, T., Kuznetsov, S. N., Klekachev, A. V., Shingubara, S. and Vanacken, J. et al. 2012. "Origin of Visible Photoluminescence from Arrays of Vertically Arranged Si-Nanopillars Decorated with Si-Nanocrystals." *Nanotechnology* 23 (Nov. 1): 475709.

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- [9] Lü, Q., Wang, J., Liang, C., Zhao, L. and Jiang, Z. 2013. "Strong Infrared Photoluminescence from Black Silicon Made with Femtosecond Laser Irradiation." *Opt. Lett.* 38 (April): 1274-6.
- [10] Ghosh, R., Giri, P. K., Imakita K. and Fujii, M. 2014. "Origin of Visible and Near-Infrared Photoluminescence from Chemically Etched Si Nanowires Decorated with Arbitrarily Shaped Si Nanocrystals." *Nanotechnology* 25 (Jan): 045703.
- [11] Fukushima, T., Ohnaka, A., Takahashi, M. and Kobayashi, H. 2011. "Fabrication of Low Reflectivity Poly-Crystalline Si Surfaces by Structure Transfer Method." *Electrochem. Solid-State Lett.* 14: B13-B15.
- [12] Imamura, K., Franco Jr., F. C., Matsumoto, T. and Kobayashi, H. 2013. "Ultra-Low Reflectivity Polycrystalline Silicon Surfaces Formed by Surface Structure Chemical Transfer Method." *Appl. Phys. Lett.* 103: 013110.
- [13] Takahashi, M., Fukushima, T., Seino, Y., Kim, W.-B., Imamura, K. and Kobayashi, H. 2013. "Surface Structure

Chemical Transfer Method for Formation of Ultralow Reflectivity Si Surfaces." *J. Electrochem. Soc.* 160 (8): H443-H445.

- [14] Brunner, R., Pinčík, E., Kobayashi, H., Kučera, M., Takahashi, M. and Rusnák, J. 2010. "On Photoluminescence Properties of a-Si:H-Based Structures." *Appl. Surf. Sci.* 256: 5591-6.
- [15] Pincik, E., Brunner, R., Kobayashi, H., Mikula, M., Vojtek, P. and Greguš, J. et al. 2016. "The Photoluminescence of Multicolor Silicon." *Journal of the Chinese Advanced Materials Society* 4 (April): 158-71.
- [16] Pincik, E., Jergel, M., Bartos, J., Falcony, C. and Kucera, M. 1999. "The C-V Investigation of Light-Related Properties of Porous Silicon/Crystalline Silicon Structure." *Superficies y Vacio* 9: 78.
- Grundman, M. 2010. The Physics of Semiconductors, An Introduction Including Nanophysics and Applications. XXXVII, 864 pages, ISBN: 978-3- 642-13 883-6, Springer-Verlag, Berlin. http://www.springer.com/978-3-642-13883-6.